

Incorporation of a Phosphonic Acid Isostere of Aspartic Acid into Peptides Using Fmoc-Solid Phase Synthesis

P. A. Lohse*# and R. Felber

Novartis Pharma Ltd., CH-4002 Basel, Switzerland

Received 10 November 1997; revised 22 December 1997; accepted 14 January 1998

Abstract: A short synthesis of a novel Fmoc-derivative 2b of the phosphonic acid isostere 1 of aspartic acid is presented. Incorporation of 2b into peptides was readily achieved using standard Fmoc-solid phase synthesis. Efficient removal of the allyl protecting groups after sequence assembly under mild conditions using Pd(0) catalysis afforded phosphonopeptides 3a and 3b in high purity. © 1998 Elsevier Science Ltd. All rights reserved.

We have been interested in exploring the role of aspartic acid isosteres in biologically active peptides. In particular we desired to study the effect of substituting the β-carboxylic acid group of aspartic acid with the more strongly acidic phosphonic acid group. A review of the literature revealed that unprotected 2-amino-3-phosphonopropionic acid (rac.-AP-3) as well as the individual enantiomers L-(R)-AP-3 (1) and D-(S)-AP-3 have been used for the synthesis of enzyme inhibitors² and simple dipeptides. Incorporation of 1 into longer peptides has not been reported. The dimethyl phosphonate 2a was proposed to be a protected form of 1 suitable for peptide synthesis. However, the use of phosphonic acid dialkylesters in peptide synthesis is limited due to the difficulties connected with cleavage of the alkyl esters after peptide assembly. In particular, incomplete diester cleavage⁵ on the one hand and instability of N-terminal formamides and carbamates on the other hand are expected to result in the formation of mixtures of products. We required a derivative of 1 which is compatible with general solid phase peptide synthesis of phosphonopeptides carrying a N-terminal carbamate-functionality.

H₂N COOH FmocHN COOH

$$O > P$$
 $O > P$
 $O >$

0040-4039/98/\$19.00 © 1998 Elsevier Science Ltd. All rights reserved. *PII*: S0040-4039(98)00189-0

[&]quot;Current address: Phylos Inc., 300 Putnam Ave., Cambridge, MA 02139, U.S.A.

In analogy to a strategy described by Shapiro et al. for the incorporation of the phosphonic acid isostere of glutamic acid into diverse peptide sequences, we chose the protected amino acid phosphonate 2b as a building block for the introduction of L-(R)-AP-3 (1) into peptides. In the following, we will describe a straightforward synthesis of 2b followed by its successful incorporation into the peptides 3a and 3b using solid phase peptide synthesis.

N-protected β -substituted L-alanines can be easily obtained in stereochemically pure form by Mitsunobu-type cyclization of readily available L-serine derivatives followed by nucleophilic ring-opening of the resulting β -lactones. Recently, Hutchinson and Parkes have shown that nucleophilic addition of dimethyl(trimethylsilyl)phosphite to 6 affords the carboxylic trimethylsilyl ester of 2a by preferential transfer of the trimethylsilyl group. We have prepared phosphite 5 by reaction of allyl alcohol with phosphorous trichloride followed by treatment with trimethylsilylchloride in the presence of triethylamine. Heating of β -lactone 6 in neat phosphite 5 followed by a simple aqueous work up yielded the diallylester 2b (Scheme).

PCI₃
$$\xrightarrow{a}$$
 \xrightarrow{b} $(CH_3)_3SiO$ \xrightarrow{b} $(CH_3)_3SiO$ \xrightarrow{b} \xrightarrow{b} $(CH_3)_3SiO$ \xrightarrow{b} \xrightarrow{b} \xrightarrow{b} \xrightarrow{b} \xrightarrow{b} \xrightarrow{c} \xrightarrow{b} \xrightarrow{c} \xrightarrow{b} \xrightarrow{c} \xrightarrow{b} \xrightarrow{c} \xrightarrow{c}

Scheme: a) 2-propenol, 58 %; b) TMSCl, NEt₃, 79%; c) 5, 100 °C, aqueous workup, 42 %. 11

Incorporation of 1 into peptides using building block 2b was demonstrated by standard Fmoc-solid phase synthesis of the two peptides Fmoc-Arg-Gly-(AP-3)-Phe-OH 3a and Ac-Arg-Gly-(AP-3)-Phe-OH 3b. These peptides are N-capped isosteres of the tetrapeptide H-Arg-Gly-Asp-Phe-OH which is known to be a potent inhibitor of fibrinogen binding to the integrin $\alpha_{IIb}\beta3$ on platelets.¹²

Starting from H-Phe-O-Wang resin, building block **2b** was attached in a single coupling using TPTU followed by Fmoc-cleavage and coupling of the next two amino acids to afford Fmoc-Arg(PMC)-Gly-[AP-3(OAllyl)₂]-Phe-O-Wang. The allyl ester protection was cleaved by stirring the resin in a solution of 2 eq. Pd(PPh₃)₄ in N-methylmorpholine/ acetic acid/ chloroform 1:2:37 for 5 h at room temperature. Subsequently, the peptide was cleaved from the solid phase using trifluoracetic acid and precipitated from ether to afford peptide **3a**. Peptide **3b** was prepared in analogy. Mass recoveries of **3a** and **3b** were good and the HPLC profiles demonstrate the high purity of the peptide products (Figure). High resolution

electrospray mass spectroscopy showed molecular parent ions in agreement with the constitution $C_{35}H_{42}N_7O_{10}P$ for **3a** and $C_{22}H_{34}N_7O_9P$ for **3b**.

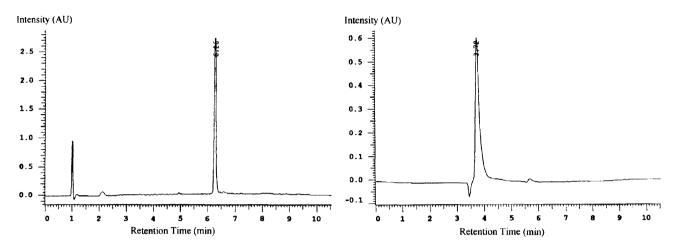


Figure: HPLC profiles of precipitated peptide products 3a (left) and 3b (right). 17

In conclusion, we have demonstrated that L-(R)-AP-3 (1), a phosphonic acid isostere of aspartic acid, can be efficiently incorporated into peptides using the building block **2b** and general Fmoc-solid phase peptide synthesis. Synthesis of **2b** is short, easy to perform and allows preparation of **2b** in gram quantities. Efficient cleavage of the phosphonate diallyl esters after peptide assembly using Pd(0) catalysis allowed us to prepare the tetrapeptides **3a** and **3b** in high purity. Our results indicate that **2b** should be well suited for the synthesis of more complicated phosphonopeptides.

Acknowledgments: We thank Drs. G. Shapiro, H. Fretz and S. Veenstra for stimulating discussion, and Mrs. G. M. D'Addio for excellent technical assistance.

References and Notes

- 1. L-2-Amino-3-phosphonopropionic acid ((L)-AP-3) was referred to as (S)-AP-3 in the literature. ^{4, 18} In contrast to this assignment and according to the Cahn-Ingold Prelog rules, ¹⁹ we think that (L)-AP-3 is identical to (R)-AP-3.
- a. Patel, D. V.; Schmidt, R. J.; Biller, S. A.; Gordon, E. M.; Robinson, S. S.; Manne, V. J. Med. Chem. 1995, 38, 2906-2921.
 b. Rosowsky, A.; Forsch, R. A.; Moran, R. G.; Kohler, W.; Freisheim, J. H. J. Med. Chem. 1988, 31, 1326-1331.
 c. Kafarski, P.; Lejczak, B.; Mastalerz, P.; Dus, D.; Radzikowski, C. J. Med. Chem. 1985, 28, 1555-1558.
- a. Nelson, V.; Mastalerz, P. J. Pharm. Sci. 1984, 73, 1844-1846. b. Kim, S. B.; Cho, S. K.; Han, J. S.; Kim, Y. J.; Hong, S.-I. J. Korean Chem. Soc. 1994, 38, 516-520.
- 4. Hutchinson, J. P. E.; Parkes, K. E. B. Tetrahedron Lett. 1992, 33, 7065-7066.
- 5. a. Otaka, A.; Burke, T. R.; Smyth, M. S.; Nomizu, M.; Roller, P. P. Tetrahedron Lett. 1993, 34, 7039-7042. b. Garbay-Jaureguiberry, C.; Ficheux, D.; Roques, B. P. Int. J. Peptide Protein Res. 1992, 39, 523-527.
- a. Kafarski, P.; Lejczak, B.; Mastalerz, P.; Szewczyk, J.; Wasielewski, C. Can. J. Chem. 1982, 60 3081-3084.
 b. Chakravarty, P. K.; Combs, P.; Roth, A.; Greenlee, W. J. Tetrahedron Lett. 1987, 28, 611. c. Sawamura, M.; Ito, Y.; Hayashi, T. Tetrahedron Lett. 1989, 30, 2247. d. Whitten, J. P.; Baron, B. M.; Muench, D.; Miller, F.; White, H. S.;

- McDonald, I. A. J. Med. Chem. 1990, 33, 2961-2963. e. Heckendorn, R.; Allgeier, H.; Baud, J.; Gunzenhauser, W.; Angst, C. J. Med. Chem. 1993, 36, 3721-3726.
- 7. Shapiro, G.; Buechler, D.; Enz, A.; Pombo-Villar, E. Tetrahedron Lett. 1994, 35, 1173-1176.
- 8. Arnold, L. D.; Kalantar, T. H.; Vederas, J. C. J. Am. Chem. Soc. 1985, 107, 7105-7109.
- 9. The dialkylphosphite 4 was prepared in analogy to the protocol for the synthesis of Bis[2,2,2-trifluoroethyl]phosphite by Gibbs, D. E.; Larsen, C. Synthesis 1984, 410-413. The crude product was distilled at 0.4 mbar and 43 °C to give pure 4 (for analytical data see: Kers, A.; Kers, I.; Stawinski, J.; Sobkowski, M.; Kraszewski, A. Synthesis 1995, 427). Synthesis of diallyl(trimethylsilyl)phosphite 5 was carried out in analogy to the procedure described: Afarinkia, K.; Rees, C. W.; Cadogan, J. I. G. Tetrahedron 1990, 46, 7187. The crude product was distilled at 0.5 mbar and 70 °C to give pure 5.
- A suspension of 1.2 g (3.9 mmol) 6^4 in 9 ml freshly distilled 5 was heated under argon at 100 °C. After stirring for 24 h, the 10. reaction had stopped although there was still a significant amount of starting material (6) left (TLC analysis on silica gel: methylene chloride/ methanol/ acetic acid 95:5:1, $R_f = 0.3$). The volatile components were removed under reduced pressure and the residue was resuspended in 9 ml of freshly distilled 5. After stirring for an additional 14 h at 100 °C the mixture was homogenous and the reaction mixture was concentrated in vacuo. The residue was taken up in ethyl acetate, washed with water, dried over sodium sulfate and purified by flash chromatography on silica gel eluting with methylene chloride/ methanol/ acetic acid (98:2:1) to give 0.77 g product 2b which crystallized upon cooling. Analytical data for 2b: H-NMR (500 MHz, DMSO-d₆): 12.66 (br., 1H, HOOC), 7.88-7.31 (8H, arom. H(Fmoc)), 7.70 (d, 1H, HN), 5.89 (m, 2H, H(olef.)), 5.30 (m, 2H, H(olef.)), 5.16 (m, 2H, H(olef.)), 4.43 (m, 4H, H₂COP), 4.29 (d, 2H, H₂C(Fmoc)), 4.25 (m, 1H, HC(α)), 4.21 (t, 1H, HC(Fmoc)), 2.30 (m, 2H, $H_2C(\beta)$); ¹³C-NMR (125 MHz, DMSO-d₆): 172.38 (s, COO, d $^3J_{CP} = 17$), 155.64 (s, OCON), 143.74 (s, Fmoc), 140.67 (s, Fmoc), 133.38 (d, olef., d ${}^{3}J_{CP} = 7$), 127.59 (d, Fmoc), 127.03 (d, Fmoc), 125.18 (d, Fmoc), 120.07 (d, Fmoc), 117.19 (t, olef., d ${}^{4}J_{CP} = 6$), 65.70 (t, Fmoc), 65.56 (t, COP, d ${}^{2}J_{CP} = 6$), 48.85 (d, C α), 46.56 (d, Fmoc), 26.41 (t, C β , d ${}^{1}J_{CP} = 141$); ${}^{31}P$ -NMR (202.5 MHz, DMSO-d₆): 29.60; MS (ESpos.): 472 (M+H)⁺; [α] ${}^{20}D = -7.4$ (c = 0.45, methanol); Mosher analysis of L-(R)-AP-3 (1, obtained by exhaustive acid hydrolysis of 2b) showed 10 % of the enantiomer D-(S)-AP-3.18
- 11. Reaction conditions were not optimized.
- 12. D'Souza, S. E.; Ginsberg, M. H.; Plow, E. F. TIBS 1991, 16, 246-250.
- 13. Fmoc-L-Phe-O-Wang resin was purchased from Novabiochem (Wang-resin: p-alkoxybenzyl-polystyrene). Fmoc cleavage was carried out with 20 % piperidine in dimethylacetamide (DMA) and resin washes with DMA-isopropanol-DMA. Building block 2b, Fmoc-Gly-OH (Bachem) and Fmoc-L-Arg(PMC)-OH (Bachem) were used for peptide-couplings. Flash-purified 2b was coevaporated with toluene to remove residual acetic acid before coupling. Couplings were performed in standard fashion in N-methylpyrrolidone using 2 eq. of Fmoc-amino acid and 2-(2-pyridon-1-yl)-1,1,3,3-tetramethyluroniumfluoroborate (TPTU, Fluka)/ diisopropylamine/ Fmoc-amino acid in a 1:1:1 ratio. Coupling was complete after 2 h reaction time as indicated by the Kaiser test: Kaiser, E. T.; Colescott, R. L.; Bossinger, C. D.; Cock, P. I. Anal. Biochem. 1970, 54, 595.
- 14. Kates, S. A.; Daniels, S. B.; Sole, N. A.; Barany, G.; Albericio, F. *Peptides, Structure & Biology*, Proc. 13 th American Peptide Symposium, ESCOM, Leiden, 1994, p. 114. Cleavage of the allyl ester protecting groups using the rapid azide-mediated, palladium (0) catalyzed cleavage works as well and yielded phosphonopeptides of comparable purity: Shapiro, G.; Buechler, D. *Tetrahedron Lett.* 1994, 35, 5421-5424.
- 15. After washing with dimethylacetamide and methylene chloride, the resin was treated with trifluoracetic acid/ water 95:5 for 3 h at room temperature. The resin was filtered off and the filtrate was added to a 20 fold excess of hexane/ tert. butyl methyl ether 1:9. After standing for 45 min. at 4 °C, the precipitate was collected by centrifugation and dried under high vacuum to yield 3a as a colorless powder.
- 16. Synthesis of **3b** started out from Fmoc-Arg(PMC)-Gly-[AP-3(OAllyl)₂]-Phe-O-Wang followed by cleavage of the Fmoc protecting group ¹³ and N-acetylation using dimethylacetamide/ acetic anhydride/ pyridine 8:1:1 for 2 min. at room temperature to give Ac-Arg(PMC)-Gly-[AP-3(OAllyl)₂]-Phe-O-Wang. Isolation of the peptide was carried out as described ¹⁵ to yield **3b** as a colorless powder.
- 17. HPLC of 3a on a 125/3 Nucleosil 120-3 C18 AB column (Marchery-Nagel): Linear gradient of H₂O/ 0.1 % trifluoracetic acid (eluent A) and acetonitrile/ 0.1 % trifluoracetic acid (eluent B) from 2 % to 100 % B over 10 min; flow rate 0.7 ml/ min, detection at 215 nm, retention time: 6.26 min. HPLC of 3b on a 250/4 Nucleosil 300-5 C4 column (Marchery-Nagel): Linear gradient of H₂O/ 0.1 % trifluoracetic acid (eluent A) and acetonitrile/ 0.1 % trifluoracetic acid (eluent B) from 2 % to 60 % B over 10 min; flow rate 0.7 ml/min, detection at 215 nm, retention time: 3.72 min.
- 18. Smith, E. C. R.; McQuaid, L. A.; Paschal, J. W.; DeHoniesto, J. J. Org. Chem. 1990, 55, 4472-4474.
- 19. Cahn, R. S.; Ingold, C. K.; Prelog, V. Angew. Chem. Int. Ed. Engl. 1966, 5, 385-415.